

## Assessment of Environmental Impact On, Wadi El-Natron Depression Lakes Water, Egypt

*M.F. Sayed and M.H. Abdo*

National Institute of Oceanography and Fisheries, Egypt

**Abstract:** Assessment of environmental Impact was established to provide and evaluated the background picture of water quality of Wadi El-Natron depression Lakes water. Physico-chemical characteristics (air and water temperatures, electrical conductivity (EC), total solids (TS), total dissolved solids (TDS), total suspended solids (TSS), salinity, pH, dissolved oxygen (DO), chemical oxygen demand (COD),  $\text{CO}_3^{2-}$ ,  $\text{HCO}_3^-$ ,  $\text{Cl}^-$ ,  $\text{SO}_4^{2-}$ ,  $\text{Na}^+$ ,  $\text{K}^+$ ,  $\text{Ca}^{2+}$ ,  $\text{Mg}^{2+}$ ,  $\text{NO}_2^-$ ,  $\text{NO}_3^-$ ,  $\text{NH}_3$ ,  $\text{PO}_4^{3-}$ , total phosphorus (TP) and  $\text{SiO}_2$ ) were measured and used as indicators to evaluate the water quality of lakes. Results indicated that the salinity ranged between 20-400‰ (hypersaline water). EC 88000-201000  $\mu\text{S}/\text{cm}$ . TS, TDS and TSS are very high levels. pH values were found to be in the alkaline side 9.12-9.84. and  $\text{HCO}_3^-$ -concentrations are very high and ranged between 3000-12000 and 1330-6480 mg/l, respectively.  $\text{SO}_4^{2-}$ -and  $\text{Cl}^-$ -are very high concentrations.  $\text{Na}^+$  and  $\text{K}^+$  are more concentrations than  $\text{Ca}^{2+}$  and  $\text{Mg}^{2+}$ . DO was near to depletion. COD is slightly high concentrations. The nutrient salts,  $\text{NO}_2^-$ ,  $\text{NO}_3^-$ -are not detected during all investigation seasons.  $\text{NH}_3$  was ranged between 0.03-0.514 mg/l.  $\text{PO}_4^{3-}$ -and TP were high concentrations during all seasons except for autumn.  $\text{SiO}_2$  was ranged between 0.7-2.25 mg/l during autumn / summer and not detected during winter / spring. It was concluded that Wadi El-Natron depression lakes are a unique aquatic ecosystem among saline lakes due to the hyper saline and alkaline waters.

**Key words:** Physico-chemical characteristics % Water % Wadi El-Natron depression lakes % Egypt

### INTRODUCTION

Inland saline lakes have received attention in recent years due to their sensitivity to climatic changes. Climatic conditions must reach a certain degree of aridity for effectively removing of water by evaporation or freeze drying and so produce a progressively concentrated brine [1,2]. Changes in evaporation rates and precipitation can affect the physical and chemical characteristics in such lakes [3-7]. Changes in water chemistry and lake depth in turn control the distribution and abundance aquatic life. In this respect, inland saline lakes need to be considered in developing concepts concerning the geochemical and biological characteristics of evaporation [8].

Wadi El-Natron, with its alkaline lakes, is an elongated depression about 90 Km North West of Cairo between latitudes  $30^\circ 15'$  north and longitude  $30^\circ 30'$  east (Fig. 1). Its average length is 60 Km and average width about is 10 Km. The bottom of the Wadi is 23 m below sea-level and 38 m below the water level of the Rosetta branch of the Nile [9]. The lowest part of the depression, encircled by contour zero, covers an area about 272  $\text{Km}^2$ . Inland lakes and crusts occupy the area surrounded by contour zero [10].

Chemical and physical studies on the Wadi El-Natron lakes are very few. Taher [8] reported that, the area of the Wadi has a mainly arid climate, with low and very variable rainfall, a long dry summer, high rates of evaporation and low humidity. All lakes had pH values of 8.5-9.5 and salinity from 283-540 g/l. The main ionic components were  $\text{SO}_4^{2-}$ ,  $\text{Cl}^-$ ,  $\text{CO}_3^{2-}$ -and  $\text{Na}^+$ . Traces of  $\text{Mg}^{2+}$  were also present.

The present study aimed to give a complete information on the physical and chemical characteristics of Wadi El-Natron depression lakes water.

### MATERIALS AND METHODS

The present study was extended from autumn 2003 to summer 2004 during four successive seasons. In the first big area lake (I) name El-Bida four brine water samples on the shore sides around the lake were collected. In the second small area lake name El-Hammra, one brine water sample was collected from the shore side of the lake. As well as ground water sample was collected from well originated on the first lake, this sample was considered as a control. The positions of stations I, II, III and IV in the first lake, station V in second lake and well ground water are represented in the map of studied (Fig. 1).



Table 1: Physico-chemical measurements of the water in the Wadi El-Natron depression lakes during autumn 2003

| Stations Parameters                               | First Lake |        |        |        | Second Lake |              |
|---|------------|--------|--------|--------|-------------|--------------|
|   | I          | II     | III    | IV     | V           | Well Control |
| Air Temperature (°C)                              | 30         | 31     | 31     | 31     | 31          | 30           |
| Water Temperature (°C)                            | 26         | 27     | 26     | 27     | 27          | 25           |
| EC (µScmG <sup>l</sup> )                          | 105300     | 88000  | 178000 | 133100 | 201000      | 700          |
| Salinity (‰)                                      | 20         | 30     | 120    | 160    | 300         | 0.0          |
| TS (mgG <sup>l</sup> )                            | 140684     | 111124 | 314948 | 208220 | 403804      | 1620         |
| TDS (mgG <sup>l</sup> )                           | 79432      | 75204  | 187300 | 137268 | 273347      | 1268         |
| TSS (mgG <sup>l</sup> )                           | 61252      | 35920  | 127648 | 70952  | 130457      | 352          |
| pH  | 9.63       | 9.71   | 9.44   | 9.55   | 9.40        | 8.24         |
| DO (mgG <sup>l</sup> )                            | 2.8        | 2.4    | 0.8    | 0.6    | nil         | 4            |
| COD (mgG <sup>l</sup> )                           | 27         | 29     | 31     | 29     | 32          | 13           |
| CO <sub>3</sub> <sup>-</sup> (mgG <sup>l</sup> )  | 4000       | 3000   | 5000   | 5400   | 9000        | 10           |
| HCO <sub>3</sub> <sup>-</sup> (mgG <sup>l</sup> ) | 1330       | 1330   | 3240   | 1940   | 4860        | 373          |
| Cl <sup>-</sup> (mgG <sup>l</sup> )               | 35500      | 30130  | 85100  | 53200  | 131200      | 171          |
| SO <sub>4</sub> <sup>-</sup> (mgG <sup>l</sup> )  | 13041      | 10400  | 20600  | 19000  | 25160       | 37           |
| Ca <sup>2+</sup> (mgG <sup>l</sup> )              | 181        | 160    | 160    | 160    | 280         | 16           |
| Mg <sup>2+</sup> (mgG <sup>l</sup> )              | 50         | 146    | 146    | 146    | 195         | 10           |
| Na <sup>+</sup> (mgG <sup>l</sup> )               | 49300      | 56000  | 94300  | 58400  | 98400       | 221          |
| K <sup>+</sup> (mgG <sup>l</sup> )                | 458        | 572.6  | 1282.6 | 641.30 | 1626.10     | 13           |
| NO <sub>2</sub> <sup>-</sup> (µgG <sup>l</sup> )  | ND         | ND     | ND     | ND     | ND          | 2.18         |
| NO <sub>3</sub> <sup>-</sup> (µgG <sup>l</sup> )  | ND         | ND     | ND     | ND     | ND          | 90           |
| NH <sub>3</sub> <sup>-</sup> (mgG <sup>l</sup> )  | 0.261      | 0.80   | 0.28   | 0.28   | 0.18        | 0.16         |
| PO <sub>4</sub> <sup>-</sup> (µgG <sup>l</sup> )  | 3575       | 1446   | 674    | 838    | 707         | 213          |
| TP (µgG <sup>l</sup> )                            | 1600       | 1550   | 1524   | 1823   | 1390        | 220          |
| SiO <sub>2</sub> <sup>-</sup> (mgG <sup>l</sup> ) | ND         | ND     | ND     | ND     | ND          | 17           |

ND: Not Detected

Table 2: Physico-chemical measurements of the water in the Wadi El-Natron depression lakes during winter 2004

| Stations Parameters                               | First Lake |        |        |        | Second Lake |              |
|---|------------|--------|--------|--------|-------------|--------------|
|   | I          | II     | III    | IV     | V           | Well Control |
| Air Temperature (°C)                              | 33         | 33     | 33     | 34     | 34          | 34           |
| Water Temperature (°C)                            | 29         | 28     | 29     | 29     | 29          | 28           |
| EC (µScmG <sup>l</sup> )                          | 164200     | 166000 | 165600 | 167000 | 184000      | 662          |
| Salinity (‰)                                      | 170        | 173    | 175    | 180    | 400         | 0.0          |
| TS (mgG <sup>l</sup> )                            | 192000     | 195000 | 196000 | 202000 | 381000      | 428          |
| TDS (mgG <sup>l</sup> )                           | 112748     | 112200 | 112000 | 119000 | 263220      | 300          |
| TSS (mgG <sup>l</sup> )                           | 79252      | 82800  | 84000  | 83000  | 117780      | 128          |
| pH  | 9.73       | 9.70   | 9.79   | 9.73   | 9.81        | 8.30         |
| DO (mgG <sup>l</sup> )                            | 1          | 0.8    | 0.7    | 0.6    | 0.5         | 1.8          |
| COD (mgG <sup>l</sup> )                           | 18         | 23     | 25     | 26     | 34          | 4            |
| CO <sub>3</sub> <sup>-</sup> (mgG <sup>l</sup> )  | 9000       | 8600   | 8800   | 10400  | 11000       | 13           |
| HCO <sub>3</sub> <sup>-</sup> (mgG <sup>l</sup> ) | 5508       | 3000   | 3300   | 2268   | 4860        | 300          |
| Cl <sup>-</sup> (mgG <sup>l</sup> )               | 116985     | 117000 | 117885 | 124075 | 233970      | 255          |
| SO <sub>4</sub> <sup>-</sup> (mgG <sup>l</sup> )  | 17700      | 18000  | 18500  | 19000  | 25000       | 36           |
| Ca <sup>2+</sup> (mgG <sup>l</sup> )              | 802        | 980    | 1100   | 1202   | 1202        | 400          |
| Mg <sup>2+</sup> (mgG <sup>l</sup> )              | 488        | 280    | 300    | 244    | 732         | 244          |
| Na <sup>+</sup> (mgG <sup>l</sup> )               | 43700      | 44000  | 44500  | 45650  | 63800       | 421          |
| K <sup>+</sup> (mgG <sup>l</sup> )                | 466        | 500    | 550    | 835    | 737         | 7.50         |
| NO <sub>2</sub> <sup>-</sup> (µgG <sup>l</sup> )  | ND         | ND     | ND     | ND     | ND          | 22           |
| NO <sub>3</sub> <sup>-</sup> (µgG <sup>l</sup> )  | ND         | ND     | ND     | ND     | ND          | 66           |
| NH <sub>3</sub> <sup>-</sup> (mgG <sup>l</sup> )  | 0.194      | 0.191  | 0.192  | 0.190  | 0.514       | 0.335        |
| PO <sub>4</sub> <sup>-</sup> (µgG <sup>l</sup> )  | 56         | 58     | 55     | 82     | 102         | 122          |
| TP (µgG <sup>l</sup> )                            | 556        | 550    | 500    | 511    | 325         | 312          |
| SiO <sub>2</sub> <sup>-</sup> (mgG <sup>l</sup> ) | 2.00       | 2.00   | 2.00   | 2.11   | 2.25        | 12.00        |

ND: Not Detected

Table 3: Physico-chemical measurements of the water in the Wadi El-Natron depression lakes during spring 2004

| Stations Parameters                                | First Lake |        |        |        | Second Lake |              |
|--|------------|--------|--------|--------|-------------|--------------|
|  | I          | II     | III    | IV     | V           | Well Control |
| Air Temperature (°C)                               | 38         | 38     | 37     | 38     | 39          | 38           |
| Water Temperature (°C)                             | 36         | 35     | 35     | 36     | 36          | 30           |
| EC (µScmG <sup>l</sup> )                           | 145000     | 146300 | 146700 | 147000 | 164200      | 604          |
| Salinity (‰)                                       | 121.2      | 131.3  | 131.3  | 130    | 414         | 0.0          |
| TS (mgIG <sup>l</sup> )                            | 142440     | 157680 | 154060 | 155790 | 428740      | 1230         |
| TDS (mgIG <sup>l</sup> )                           | 115480     | 146880 | 148610 | 145140 | 383360      | 1100         |
| TSS (mgIG <sup>l</sup> )                           | 26960      | 10800  | 5450   | 10650  | 4510        | 130          |
| pH   | 9.26       | 9.28   | 9.29   | 9.30   | 9.12        | 7.97         |
| DO (mgIG <sup>l</sup> )                            | 0.53       | 0.47   | 0.33   | 0.55   | 0.27        | 4.00         |
| COD (mgIG <sup>l</sup> )                           | 26         | 24     | 23     | 25     | 25          | 8            |
| CO <sub>3</sub> <sup>-</sup> (mgIG <sup>l</sup> )  | 4600       | 5000   | 5100   | 5200   | 11000       | 13           |
| HCO <sub>3</sub> <sup>-</sup> (mgIG <sup>l</sup> ) | 2536       | 2592   | 2268   | 2592   | 6480        | 381          |
| Cl <sup>-</sup> (mgIG <sup>l</sup> )               | 63810      | 71000  | 71000  | 74500  | 173700      | 128          |
| SO <sub>4</sub> <sup>-</sup> (mgIG <sup>l</sup> )  | 3840       | 6320   | 5900   | 6630   | 8250        | 126          |
| Ca <sup>2+</sup> (mgIG <sup>l</sup> )              | 88         | 66     | 66     | 66     | 56          | 16           |
| Mg <sup>2+</sup> (mgIG <sup>l</sup> )              | 40         | 34     | 34     | 27     | 25          | 6            |
| Na <sup>+</sup> (mgIG <sup>l</sup> )               | 28300      | 29200  | 29400  | 30000  | 47600       | 173          |
| K <sup>+</sup> (mgIG <sup>l</sup> )                | 583        | 603    | 618    | 618    | 2751        | 5            |
| NO <sub>2</sub> <sup>-</sup> (µgIG <sup>l</sup> )  | ND         | ND     | ND     | ND     | ND          | 2.18         |
| NO <sub>3</sub> <sup>-</sup> (µgIG <sup>l</sup> )  | ND         | ND     | ND     | ND     | ND          | 104          |
| NH <sub>3</sub> <sup>-</sup> (mgIG <sup>l</sup> )  | 0.10       | 0.12   | 0.25   | 0.17   | 0.21        | 0.16         |
| PO <sub>4</sub> <sup>-</sup> (µgIG <sup>l</sup> )  | 2624       | 2902   | 1492   | 1621   | 416         | 996          |
| TP (µgIG <sup>l</sup> )                            | 2684       | 2973   | 2330   | 2422   | 500         | 1018         |
| SiO <sub>2</sub> <sup>-</sup> (mgIG <sup>l</sup> ) | ND         | ND     | ND     | ND     | ND          | 11.1         |

ND: Not Detected

Table 4: Physico-chemical measurements of the water in the Wadi El-Natron depression lakes during summer 2004

| Stations Parameters                                | First Lake |        |        |        | Second Lake |              |
|--|------------|--------|--------|--------|-------------|--------------|
|  | I          | II     | III    | IV     | V           | Well Control |
| Air Temperature (°C)                               | 37         | 40     | 40     | 41     | 41          | 37           |
| Water Temperature (°C)                             | 35         | 39     | 39     | 39     | 39          | 32           |
| EC (µScmG <sup>l</sup> )                           | 164500     | 163500 | 163500 | 164100 | 192500      | 930          |
| Salinity (‰)                                       | 152        | 172    | 172    | 172    | 384         | 0.2          |
| TS (mgIG <sup>l</sup> )                            | 218430     | 218350 | 198440 | 218300 | 471640      | 956          |
| TDS (mgIG <sup>l</sup> )                           | 214000     | 214120 | 194460 | 214700 | 464310      | 660          |
| TSS (mgIG <sup>l</sup> )                           | 4430       | 4230   | 3980   | 3600   | 7330        | 296          |
| pH   | 9.72       | 9.80   | 9.81   | 9.84   | 9.55        | 8.27         |
| DO (mgIG <sup>l</sup> )                            | 0.4        | 0.4    | 0.4    | 0.4    | 0.4         | 3.0          |
| COD (mgIG <sup>l</sup> )                           | 24         | 24     | 28     | 26     | 29          | 4            |
| CO <sub>3</sub> <sup>-</sup> (mgIG <sup>l</sup> )  | 10000      | 10000  | 10100  | 10200  | 12000       | 25           |
| HCO <sub>3</sub> <sup>-</sup> (mgIG <sup>l</sup> ) | 3240       | 1296   | 2916   | 3240   | 3240        | 365          |
| Cl <sup>-</sup> (mgIG <sup>l</sup> )               | 85080      | 80826  | 82953  | 81535  | 177250      | 156          |
| SO <sub>4</sub> <sup>-</sup> (mgIG <sup>l</sup> )  | 16393      | 18031  | 17506  | 17544  | 22195       | 56.00        |
| Ca <sup>2+</sup> (mgIG <sup>l</sup> )              | 46         | 40     | 40     | 40     | 40          | 15           |
| Mg <sup>2+</sup> (mgIG <sup>l</sup> )              | 49         | 73     | 73     | 73     | 78          | 12           |
| Na <sup>+</sup> (mgIG <sup>l</sup> )               | 168350     | 175670 | 176900 | 176480 | 249680      | 1132         |
| K <sup>+</sup> (mgIG <sup>l</sup> )                | 598        | 544    | 544    | 536    | 3991        | 5            |
| NO <sub>2</sub> <sup>-</sup> (µgIG <sup>l</sup> )  | ND         | ND     | ND     | ND     | ND          | 2.18         |
| NO <sub>3</sub> <sup>-</sup> (µgIG <sup>l</sup> )  | ND         | ND     | ND     | ND     | ND          | 20.10        |
| NH <sub>3</sub> <sup>-</sup> (mgIG <sup>l</sup> )  | 0.10       | 0.08   | 0.061  | 0.043  | 0.030       | 0.12         |
| PO <sub>4</sub> <sup>-</sup> (µgIG <sup>l</sup> )  | 348        | 348    | 634    | 501    | 164         | 156.4        |
| TP (µgIG <sup>l</sup> )                            | 3108       | 3223   | 2400   | 2189   | 393         | 380          |
| SiO <sub>2</sub> <sup>-</sup> (mgIG <sup>l</sup> ) | 1.20       | 1.23   | 1.55   | 1.50   | 0.70        | 11           |

ND: Not Detected

volume by > 60 % following evaporation in summer. Morphological evidence for former higher shorelines in the form of small beach scraps can be observed in some small lakes. Due to high evaporation rates in these lakes, the number of lakes present at any time is therefore variable; for instance Stocker, [13] recorded 16 lakes in Wadi, while Imhoff *et al.* [14] noted 6 lakes and Taher, [8] was collected samples from 3 lakes, in present study (2003-2004), samples were collected from only 2 lakes.

Temperature is strong and great important factor for aquatic ecosystem, as it affects the organisms as well as physical and chemical characteristics of water [15]. As expect the high water temperature was recorded during summer and the lower values in winter due to the more or less of air temperature. The area of Wadi El-Natron has a mainly arid climate with low and very variable rainfall, a long day summer, high evaporation rates and low humidity [8].

The water is undergoing significant fluctuations in EC. The EC values were recorded very high values during all seasons especially summer following by autumn then spring and the lower values in winter. The increase in EC values at all investigated stations is related to the increase in total dissolved solids and water temperatures [16]. The EC values of ground well water control were very small when compared with EC values of Wadi El-Natron lakes water. Also, values of EC at all stations of first lake were low as compared with second lake, this may be attributed to the dilution by large amount of ground water discharged from the well, situated at this lake.

Salinity is among the most important environmental factor and exerts various effects on the vitality of marine organisms. The high values of salinity were recorded during autumn, summer then spring and the lower values during winter. This could be mainly attributed to the changes in water levels of the lakes, as well as evaporation rates during different seasons. The salinity lower values of first than second lake is related to the dilution effect by the ground well water at first lake. Generally, the very high values of the Wadi El-Natron lakes water salinity, could be mainly attributed to the salty rock nature of the Natrun area. Therefore water of these lakes is classified under hyper saline water (EC, greater than 60000  $\mu\text{S}/\text{cm}$ ) according to Bowman [17].

Solids refer to suspended and dissolved matter in water. They very useful parameter describing the chemical constituents of the water and can be considered as general of edaphic relations that contribute to productivity within the water body [16]. TS and TDS values were found in the same trend of the EC and

salinity. However, the higher values of TS and TDS were recorded during summer 198440-218430 and 194460-214700 mg/l, respectively in first lake and it was of very high values in second lake 471640 and 464310 mg/l, respectively, this may be due to dilution effect of ground water in first lake. On the other side, the lower values were recorded during winter. This also could be due to the high evaporation rates during summer which facilitate in the accumulation of different dissolved salts in Natrun Lake water. TS and TDS ranges of ground well water control were found to be 425-1620 and 300-1268 mg/l, respectively during different season. This results were coincident with that reported by Taher [8] on the same lakes. He found that TDS ranged between 283000-557000 mg/l in first and second lakes.

Wadi El-Natron lakes water classified undergoing highly alkaline water during all investigation seasons without spatial variation. The ranges of pH values in first lake were 9.26-9.85 and in second lake 9.12-9.81 during different seasons. On the other side, pH values of ground water control were found to be 7.97-8.30. This result agrees with that found by Taher [8]. He recorded that, lake water was highly alkaline and pH was reasonably constant ranging from 8.51 in first lake and 9.45 in second lakes.

Wadi El-Natron lakes water is very poor in oxygen during all investigation seasons. The ranges were found to be 0.33-2.8 and nil-0.5 mg/l in first and second lakes, respectively during all seasons. Also, the ground water is defined and characterized by very low DO. However DO ranging from 1.8-4 mg/l in the current well water. The very low DO near to depletion may be attributed to the nature of the close and hyper saline lakes, but this lead to death of most phyto and zooplankton species as well as the lakes are dead and become similar to Dead Sea. BOD was not available to determined because the very low DO. Frequent occurrences of hypoxia due to sudden shutdown of DO have caused significant reduction of fishery harvests, toxic algal blooms and biotic diversity [18] and may result in reducing conditions in sediments, which may release previously bound toxicants into the water column [19]. The recorded values of DO at all stations of first and second lakes were less than the guideline values cited by USEPA [20] for the protection of aquatic life (for warm water biota: early life stages = 6 mg/l, other life stages = 5.5 mg/l for cold water biota; early life stages = 9.5 mg/l, other life stage = 6.5 mg/l).

COD values were inverse to DO and were significant during all investigation seasons. The significant values of COD may be due to organic acids originating from

decaying swamp grass in the depression and microbial decomposition of organic matter in the sediment of the lakes. This interpretation is coincident with that reported by Taher [8]. The high values of pH is an indication to the high values of  $\text{CO}_3^-$  and  $\text{HCO}_3^-$  of Wadi El-Natron lakes water during different seasons. The maximum concentrations of  $\text{CO}_3^-$  were recorded during summer 10000-12000 mg/l and the lower values were recorded during winter 3000-9000 mg/l.

The  $\text{HCO}_3^-$  concentrations showed slight variation during different seasons and ranged between 1330-6480 mg/l at two studied lakes. On the other side, the ranges of  $\text{CO}_3^-$  and  $\text{HCO}_3^-$  in the ground water control were 10-25 and 300-381 mg/l during different seasons respectively. The high concentrations of  $\text{CO}_3^-$  and  $\text{HCO}_3^-$  in Wadi El-Natron lakes water may be related to the nature of the depression area and the increase in pH values facilitate in the formation of two ions. Taher [8] determined  $\text{CO}_3^-$  and  $\text{HCO}_3^-$  and found the same ranges of the present results.

The high salinity and TDS values of Wadi El-Natron lakes water indicating that the increase in Cl- and  $\text{SO}_4^-$  concentration. However the ranges of two ions in the second were higher than in the first lake during different seasons. The values of Cl- and  $\text{SO}_4^-$  in ground water well were lower than in hyper saline water lakes. This could be mainly attributed to the weathering of gypseous clay component in Wadi El-Natron depression lake.

The calculations of the % of each anion revealed that, the major anion in Wadi El-Natron lakes water was Cl- and  $\text{SO}_4^-$ , while the minor anions  $\text{CO}_3^-$  and  $\text{HCO}_3^-$ . This results agrees with that reported by Taher [8]. He showed that, the water type of the Wadi El-Natron lakes is Cl- to  $\text{SO}_4^-$ -Cl.

$\text{Ca}^{2+}$  and  $\text{Mg}^{2+}$  concentrations were increased during autumn season and relative increase during winter. The lower values were recorded during hot seasons (spring-summer). The high values of  $\text{Ca}^{2+}$  and  $\text{Mg}^{2+}$  during cold season (autumn-winter), may be related to dissolution of  $\text{Ca}^{2+}$  and  $\text{Mg}^{2+}$  from the surrounding gypseous rocks and sediment which release into water because of the wind velocity ranges between 12 and 20 Km/h in December and June respectively [8,16].

The high salinity values and the nature of Wadi El-Natron lakes water responsible for the very high concentrations of sodium in first and second lakes during all seasons. The very high sodium concentrations were recorded during summer is mainly related to the high evaporation rates during this season which concentrated some ions especially sodium. Also, the second lake has highly salts content than first lake (diluted by ground

water discharging from well present here). The potassium concentrations behave the same trends as  $\text{Na}^+$  during all seasons.

The calculations of each cation revealed that, the  $\text{Na}^+$  is predominant cation (98.4 %) and the other cations  $\text{K}^+$ ,  $\text{Ca}^{2+}$  and  $\text{Mg}^{2+}$  are very minor. Also, this result is coincident with that reported by Taher [8] the Natrun lakes water dominated by  $\text{Na}^+$ , while  $\text{Ca}^{2+}$  and mostly  $\text{Mg}^{2+}$  were below detection limits.

Nutrient salts ( $\text{NO}_2^-$ ,  $\text{NO}_3^-$ ,  $\text{NH}_3$ ,  $\text{PO}_4^{3-}$ , TP and  $\text{SiO}_2$ ) play an important role in the productivity of the aquatic ecosystem supporting the food chain for phyto- and zooplanktons as well as fish [21].

$\text{NO}_2^-$ -N and  $\text{NO}_3^-$ -N were not detected in the Wadi El-Natron lakes water at all stations during different seasons as represented. The depletion in  $\text{NO}_2^-$  and  $\text{NO}_3^-$ , may be related to denitrification of ions into ammonia which it only detected nitrogen ion form. This interpretation is supported by Diaz *et al.* [22] for Lake Lasalada De Chiprana (Spain); Morales *et al.* [23] for Lake Maracaibo (Venezuela). On the other side,  $\text{NO}_2^-$ -N and  $\text{NO}_3^-$ -N were detected during different seasons in the ground water sample collected from the well situated at this lakes.

Ammonia-N accounted for the major proportion of total soluble inorganic nitrogen. It fluctuated in wide ranges during different seasons and within normal guide lines (0.48-2.68 mg/l at pH 8.6), USEPA [20].

The cycling of phosphorus within lakes and rivers is dynamic and complex, involving adsorption and precipitation reactions, inter change with sediments and uptake by aquatic biota [24]. The increase in the ortho and total phosphorus may be related to the high alkalinity noticeable in the Natrun lakes water which facilitate in the sedimentation of phosphorus then redissolution from sediment to water under high pH values of the lakes. And supported by Mesnage and Picot [25] and Herrera-Silvenia [26]. Also, the fluctuations of total phosphorus are mainly affected by two factors; the sedimentation rate and renewable rate of water [27], as well as the release of phosphorus from sediment under favorable conditions by either deposition or resuspension process [28].

The silica were not detected during winter / spring period, while detected during autumn / summer period. On the other hand, there is noticeable increase in silicate concentrations in ground water sample collected during different seasons which is especial character for ground water as reported by Abdo [29]. The relative increase and detection of  $\text{SiO}_2$  during autumn / summer period may be related to the alkaline pH of water accelerates the release of silicate from sediments to the over lying water [30].

Correlation coefficient ( $r$ ) is one of the most important statistical tests used for identification of correlation among different physical and chemical parameters measurements in the water ecosystem. In the present study, the EC values possessed a strong positive relationship with salinity, TS, TDS at  $P < 0.05$  ( $r = 0.91, 0.99$  and  $0.95$ ) during different seasons. Also, salinity values are positive correlation with TS and TDS ( $r = 0.9$ ) at  $P < 0.05$  during different seasons. This indication on the strong relationships among physical parameters (EC, salinity, TS and TDS) in the hyper saline ecosystem. pH values with  $\text{CO}_3^{2-}$  and  $\text{HCO}_3^-$  are dependent on each other through ( $r = 0.5, 0.37, -0.95, -0.92, -0.96, -0.98$  and  $-0.89, -0.33$ ) during autumn, winter, spring and summer seasons respectively.  $\text{CO}_3^{2-}$  concentrations had strong positive correlations with  $\text{Na}^+$  and  $\text{K}^+$  during different seasons ( $r = 0.88$  and  $0.92$ ). Also,  $\text{CO}_3^{2-}$  concentration values were found positive correlations with  $\text{Ca}^{2+}$  and  $\text{Mg}^{2+}$  ( $r = 0.71, 0.56, 0.88, 0.63$  and  $0.43$ ) during different seasons. As well as, Cl-values had strong positive correlations with  $\text{Na}^+$  and  $\text{K}^+$  ( $r = 0.98$  and  $0.86$ ), this reveals that the  $\text{Ca}^{2+}$  and  $\text{Mg}^{2+}$ ,  $\text{Na}^+$  and  $\text{K}^+$  may be occur in the Wadi El-Natron Lakes as carbonate and chloride salts. Cl-correlated significantly with  $\text{SO}_4^{2-}$  ( $r = 0.85$ ) and represented the same behaviour of two anions in Wadi El-Natron lakes.  $\text{Na}^+$  and  $\text{K}^+$  are take the same manner in the ecosystem ( $r = 0.9$ ). Nutrient salts ( $\text{NO}_2^-$ ,  $\text{NO}_3^-$  and  $\text{NH}_3$ ,  $\text{PO}_4^{3-}$ , TP and  $\text{SiO}_2$ ) had very weak in correlations.

The comparison between present results with previous studies on the same lakes, [8]. pH values in the same rages. TDS,  $\text{CO}_3^{2-}$ ,  $\text{Na}^+$  and  $\text{SO}_4^{2-}$  are lower, but  $\text{HCO}_3^-$ , Cl<sup>-</sup>,  $\text{Ca}^{2+}$  and  $\text{Mg}^{2+}$  are higher than that reported by Taher, [8]. On the other side, the comparison of the present results with another world saline lakes e.g in Alberta and Saskatchewan; pH (9.43-9.53), DO (0.31-.5 mg/l), salinity (22-160 ppt) TDS (24-161 g/l) and EC (15.6-201.35 mS/cm) determined by Bowman and Pschs [17]. These values are lower than present results except for DO are in same ranges. Also, a comparison of the Wadi El-Natron Lakes with the lakes studied by Seaman *et al.* [48] in Southern Africa revealed an entirely different chemistry. The Southern Africa lakes are much less saline (< 50 g/l), pH (9.2-10.4), the most common pattern of anion dominance is Cl<sup>-</sup> >  $\text{HCO}_3^-$  >  $\text{CO}_3^{2-}$  which is similar to Wadi El-Natron brines.

In conclusion, Wadi El-Natron appear to be a unique aquatic ecosystem among saline lakes characterized by hyper saline, highly alkaline, poor in DO,  $\text{NO}_2^-$ ,  $\text{NO}_3^-$ ,  $\text{SiO}_2$  and decreased  $\text{Ca}^{2+}$ ,  $\text{Mg}^{2+}$ ,  $\text{K}^+$  and rich in Cl<sup>-</sup>,  $\text{SO}_4^{2-}$ ,  $\text{Na}^+$ ,  $\text{CO}_3^{2-}$ ,  $\text{HCO}_3^-$  as well as relative increase in  $\text{PO}_4^{3-}$  and TP. It is recommended to increase the numbers of under

ground water wells influx in the lakes to improve the water quality and dilution of salinity lakes. As well as, these lakes in this area with these clear climatic conditions, the alkaline water of the lakes can be uses in healthy tourism purposes.

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