

## Nitrate by Nanofiltrations in Very Small Drinking Water Systems

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**Abstract:** Groundwater is an essential drinking water source in gyeongbuk, Korea. Population of water consumption has exposed to NO<sub>3</sub>-N which is one of the important water quality parameters for a long time. NO<sub>3</sub>-N concentration is varied with the season change and is increased with the passage of time. To evaluate the availability of nanofiltration membrane (NF) systems, the NO<sub>3</sub>-N removal by UTC-70UB and UTC-20 was performed. In both UTC-70UB and UTC-20, NO<sub>3</sub>-N removal is affected strongly by groundwater characteristics. UTC-70UB is a suitable system for the removal of a wide range of NO<sub>3</sub>-N from groundwaters in a view of the drinking water production. For the application of NF to small drinking water treatment systems, the water supply law must pay due regard to the state of rural areas.

**Key words:** Groundwater • Nitrate • Nanofiltration • Drinking water • Rural area

### INTRODUCTION

Nitrate concentration in water resources, particularly in groundwater, has increased to a level that potentially causes a public health threat in many regions of the world [1-3]. Recently a very big problem caused by NO<sub>3</sub>-N in small drinking water plants (small drinking water plant: water production is lesser than 10<sup>4</sup> m<sup>3</sup>/day) with groundwaters has occurred in Korea.

In the case of small drinking water plants, the quality, quantity and availability of drinking water are one of the most important and social issues at rural areas. Monitoring of water quality on the basis of data is necessary to supply safe drinking waters. The regulatory limit for nitrate (as nitrogen) in drinking water proposed by KEM(The Korean Environmental Ministry) as 10 mg/L. Even when the concentration of nitrate nitrogen was detected over the drinking water quality standard, the residents of that country have no chance to change the drinking water immediately. KEM began to improve small drinking water plants for providing the safe drinking water actively. Since groundwater is associated with an agricultural irrigation, its quality becomes worse with each passing day. Especially groundwater is polluted by nitrates responding a risk for human health [4-6]. The studied area, located in gyeongbuk, Korea, encompasses irrigated agricultural fields.

NF has shown its effectiveness in the removal of a great variety of undesirable components from water [7-8]. Its separation mechanisms combine sieving effect, differences in diffusivity and solubility of solutes and electrostatic interactions between the membrane surface groups and the ions. In the case of negatively charged membranes, anions like nitrates can be effectively removed [9]. A suitable NF membrane for nitrate removal must have tight porous structures and be negatively charged.

The present work studied the rejection of the UTC-70UB and UTC-20 with synthetic water and groundwater. The influence of nitrate rejection and groundwater quality was studied using nitrate concentration changes.

### The following points were taken into account:

- Investigation of nitrate pollution level in groundwater;
- Concentration effect on nitrate removal;
- NF feasibility in very small drinking water systems.

### MATERIALS AND METHODS

Taking contamination levels of nitrate into consideration, one small drinking water plant using

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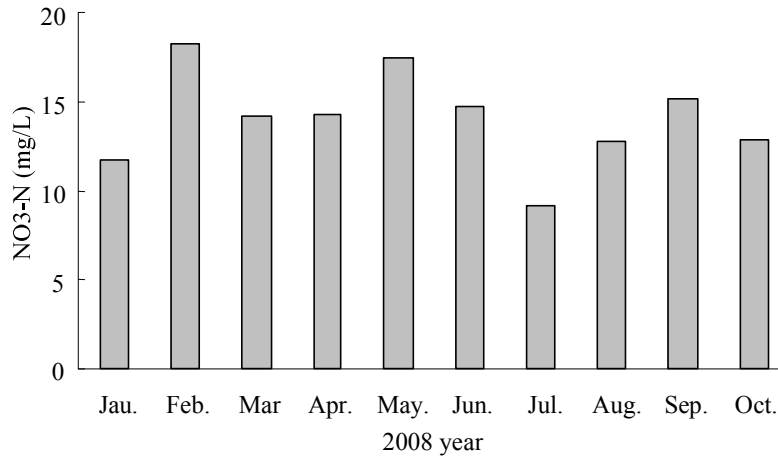


Fig. 2: NO<sub>3</sub>-N concentration in groundwater A (GWA) from Jan. to Oct., 2008

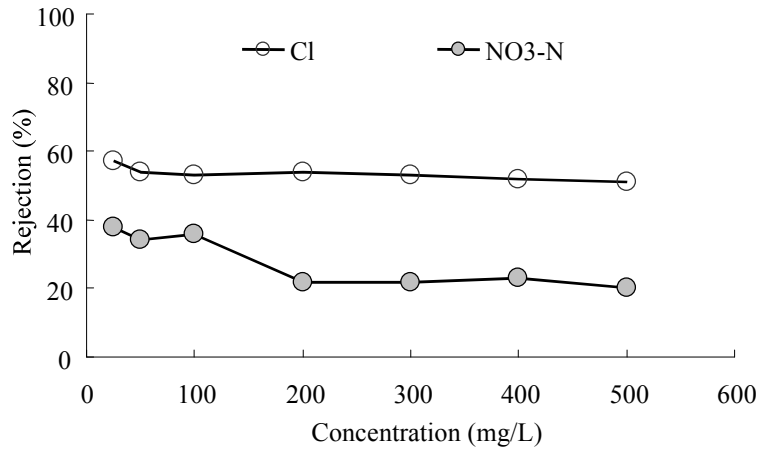


Fig. 3: Rejection of chloride ion and NO<sub>3</sub>-N by UTC-20 as a function of concentration

Table 1: Effect of NO<sub>3</sub>-N concentration on flux and rejection

NF	Flux and Rejection	Distilled water	NO <sub>3</sub> -N concentration (mg/L)		
			25	50	500
UTC-70UB	Flux (m <sup>3</sup> /day)	0.0026	0.0026	0.0025	0.0023
	Rejection (%)	-	88	88	87
UTC-20	Flux (m <sup>3</sup> /day)	0.0094	0.0079	0.0080	0.0078
	Rejection (%)	-	38	34	20

Where, rejection (%) = (feed conc.-permeated conc.)/feed conc. \*100

Table 2: Characteristics of groundwater quality

Items	Unit	GWA	GWB	GWC	GWD	GWE
pH	-	7.6	7.9	8.0	8.2	8.2
E.C.	uS/cm	477	542	255	493	943
Cl <sup>-</sup>	mg/L	25	18	8.2	60	11
SO <sub>4</sub> <sup>2-</sup>	mg/L	38	122	20	31	3.2
NO <sub>3</sub> -N	mg/L	18.3	2.0	1.2	0.3	3.3
TDS	mg/L	230	263	122	239	463
Hardness	mg/L	210	312	92	178	347

decreased with the increase of NO<sub>3</sub>-N concentrations, while that by UTC-20 is decreased with the increase of NO<sub>3</sub>-N concentrations when NO<sub>3</sub>-N concentrations are varied from 25 to 500 mg/L.

Fig.3 describes the comparison of rejection between chloride ion and NO<sub>3</sub>-N by UTC-20.

As illustrated in Fig.3, NO<sub>3</sub>-N rejection is affected by concentration levels while the chloride ion is rejected with a stable efficiency and high rejection rats. It means that NO<sub>3</sub>-N is not a simple pollutant when NO<sub>3</sub>-N is detected with a high concentration in groundwater. If the NO<sub>3</sub>-N rejection by UTC-20 is 38%, it can be achieved the drinking water standard of NO<sub>3</sub>-N with 9.92 mg/L when NO<sub>3</sub>-N concentration is lower than 16 mg/L in the feed water.

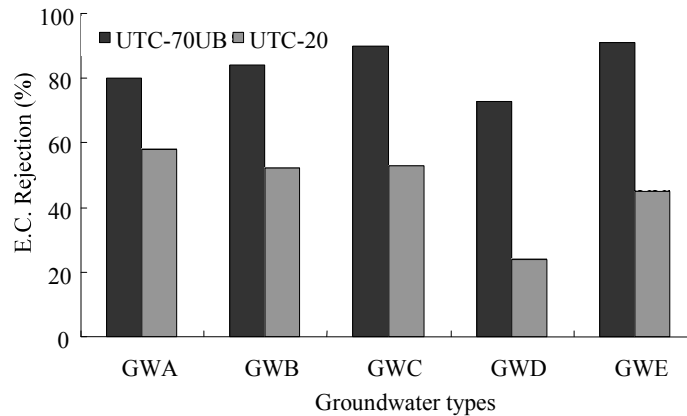


Fig. 4: Effect of groundwater quality on rejection efficiency

**Effect of Groundwater Quality on Membrane Performance:** To understand water quality effect on membrane performance, the quality of groundwater is measured as shown in Table 2. The population health is exposed by groundwater used for drinking directly. The effect of groundwater qualities on the rejection efficiency by nanofiltration as a function of electric conductivity (E.C.) is shown in Fig. 4. Five types of groundwater are described with GRA, GRB, GRC, GRD and GRE which are used for drinking water without any treatment. As shown in Fig.4, E.C. rejection is very strongly affected by the water quality, especially chloride ion. In the case of GWD, chloride ion is detected much higher than that in other groundwater. Concentration of chloride ion in GWD is 59.8 mg/L while that in other GWs is lower than 20 mg/L.

For UTC-70UB, the rejection of E.C. in GWA, GWB, GWC and GWE is higher than 80% while that in GWD is just 73%. For UTC-20, rejection in GWA, GWB and GWC is higher than 50% while that in GWD is just 24%. Also E.C. rejection is not varied directly as the nanofiltration efficiency in an actual condition of groundwaters. It can be understood according to E.C. rejection in the case of GWE. For GWE, E.C. rejection by UTC-20 shows higher than that of GWD while E.C. rejection by UTC-70B shows highest efficiency. It is demonstrated that NF system with the lower rejection efficiency is required the characteristic of water qualities more carefully to product safe drinking water.

## CONCLUSIONS

Recently  $\text{NO}_3\text{-N}$  is detected high levels which are harmful to human in many small drinking water plants in

Korea. Once  $\text{NO}_3\text{-N}$  is occurred with a high concentration, the concentration of  $\text{NO}_3\text{-N}$  is rarely reduced. The effect of  $\text{NO}_3\text{-N}$  concentration is significant in the case of UTC-20 while UTC-70UB efficiency on  $\text{NO}_3\text{-N}$  removal is not affected by  $\text{NO}_3\text{-N}$  concentration with the range of 25 to 500 mg/L. In both UTC-70UB and UTC-20, their efficiency of E.C. rejection is strongly affected by groundwater characteristics. However it is demonstrated that UTC-70UB is a suitable system for remove  $\text{NO}_3\text{-N}$  in groundwaters because  $\text{NO}_3\text{-N}$  removal efficiency is high and stable regardless of groundwater qualities.

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## REFERENCES

1. Green, M. and G. Shelef, 1994. Treatment of Nitrate-contaminated Groundwater for Drinking Purposes. In: Zoller, U. (Ed.), Groundwater Contamination and Control, Marcel Dekker, Inc, New York, 1994, pp: 587-606.
2. Kapoor, A. and T. Viraraghavan, 1997. Nitrate Removal from Drinking Water-review, J. Environ. Engin., 123(4): 371-380.
3. Moore, K., B. Ekwurel, B.K. Esser, G.B. Hudson and J.E. Moran, 2006. Sources of Groundwater Nitrate Revealed Using Residence Time and Isotope Methods, Appl. Geochem., Elsevier, 21(6): 1016-1029.

4. Moreno, B., M.A. Gomez, A. Ramos, J. Gonzalez-Lopez and E. Hontoria, 2005. Influence of Inocula Over Start up of a Denitrifying Submerged Filter Applied to Nitrate Contaminated Groundwater Treatment, *J. Hazardous Materials*, 127(1-3): 180-186.
5. U.S. Environmental Protection Agency (EPA), 1993. Wellhead Protection: a Guide for Small Communities, Office of Research and Development Office of Water, Washington, DC, (EPA/625/R-93/002).
6. World Health Organization (WHO), 1998. Guidelines for Drinking Water Quality, World Health Organization, Geneva. 1998.
7. Van der Bruggen, B. and C. Vandecasteele, **0000**. Removal of Pollutants From Surface Water and Groundwater by Nanofiltration: Overview of Possible Applications in the Drinking Water Industry, *Environ. Pollution*, 122(3): 435-445.
8. Santafe-Moros, A., J.M. Gozálvez-Zafrilla and J. Lora-Garc, 2007. Nitrate Removal from Ternary Ionic Solutions by a Tight Nonafiltration Membrane, *Desalination*, 204(1-3): 63-71.
9. Sato Yuko, Meea Kang, Tasuku Kamei and Yasumoto Magara, 2002. Performance of nanofiltration for arsenic removal, *Water Res.*, 36: 3371-3377.